

## Changes in Market Equilibrium

| Product                | Headline/Event  | How are the following affected?   | Graph |
|------------------------|---|---|-------|
| 1. Vacation trips      | "America goes back into a recession. Millions are unemployed"   | <p><u>Circle one for each.</u></p> <p>1. Affects: <i>Supply</i> OR <i>Demand</i>?<br/>                     a. <i>Decreases</i> OR <i>Increases</i>?<br/>                     b. <i>Left</i> OR <i>Right</i>?</p> <p>2. Equilibrium Price: <i>Decrease</i> OR <i>Increase</i>?</p> <p>3. Equilibrium Quantity: <i>Decrease</i> OR <i>Increase</i>?</p> <p>4. Reason/Determinant?</p> |       |
| 2. Gasoline            | "Engineers have invented a process for refining crude oil into gasoline that doubles the production of gasoline per barrel of crude oil." | <p><u>Circle one for each.</u></p> <p>1. Affects: <i>Supply</i> OR <i>Demand</i>?<br/>                     a. <i>Decreases</i> OR <i>Increases</i>?<br/>                     b. <i>Left</i> OR <i>Right</i>?</p> <p>2. Equilibrium Price: <i>Decrease</i> OR <i>Increase</i>?</p> <p>3. Equilibrium Quantity: <i>Decrease</i> OR <i>Increase</i>?</p> <p>4. Reason/Determinant?</p> |       |
| 3. Video Games         | "It becomes known that an electronics store is going to raise the price of all their video games 3 months from now."                      | <p><u>Circle one for each.</u></p> <p>1. Affects: <i>Supply</i> OR <i>Demand</i>?<br/>                     a. <i>Decreases</i> OR <i>Increases</i>?<br/>                     b. <i>Left</i> OR <i>Right</i>?</p> <p>2. Equilibrium Price: <i>Decrease</i> OR <i>Increase</i>?</p> <p>3. Equilibrium Quantity: <i>Decrease</i> OR <i>Increase</i>?</p> <p>4. Reason/Determinant?</p> |       |
| 4. Apartments for Rent | "The price for New Homes skyrockets-highest in the last seven years"  | <p><u>Circle one for each.</u></p> <p>1. Affects: <i>Supply</i> OR <i>Demand</i>?<br/>                     a. <i>Decreases</i> OR <i>Increases</i>?<br/>                     b. <i>Left</i> OR <i>Right</i>?</p> <p>2. Equilibrium Price: <i>Decrease</i> OR <i>Increase</i>?</p> <p>3. Equilibrium Quantity: <i>Decrease</i> OR <i>Increase</i>?</p> <p>4. Reason/Determinant?</p> |       |
| 5. Soy Beans           | "Congress Approves a Huge Subsidy for Soy Bean Farmers to Increase Production on Farms"   | <p><u>Circle one for each.</u></p> <p>1. Affects: <i>Supply</i> OR <i>Demand</i>?<br/>                     a. <i>Decreases</i> OR <i>Increases</i>?<br/>                     b. <i>Left</i> OR <i>Right</i>?</p> <p>2. Equilibrium Price: <i>Decrease</i> OR <i>Increase</i>?</p> <p>3. Equilibrium Quantity: <i>Decrease</i> OR <i>Increase</i>?</p> <p>4. Reason/Determinant?</p> |       |

# Economics Lesson Handout Answers

**Lingjun Ying**



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web the primary approach is to analyse and study the grammatical rules of the language usually in an order roughly matching the traditional order of the grammar of latin then practise manipulating grammatical structures through translation into and from the mother tongue the method is based on the written word and texts are widely available

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web the rate law for a particular reaction is rate  $k \times [x]^2$  in an experiment the initial rate of the reaction is determined to be  $0.080 \text{ mol l}^{-1} \text{ s}^{-1}$  when the initial concentration of x is

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web a rate law shows how the rate of a chemical reaction depends on reactant concentration for a reaction such as  $a \rightarrow \text{products}$  the rate law generally has the form rate  $k [a]^n$

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web the reaction orders in a rate law describe the mathematical dependence of the rate on reactant concentrations referring

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web each reactant in the rate limiting step of a reaction is assigned an order typically zeroth 1st or 2nd the order is the sum of all individual orders the rate of a reaction is

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web expert answer transcribed image text 2 determine the reaction orders and the rate law for the experiment determine the rate constant at room temperature 3 explain how the

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web the rate law or differential rate law relates the rate of a reaction to the concentration or pressure of the reactants the rate of a reaction is proportional to the concentration or

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web oct 27 2022 the rate law for the reaction ce h<sub>2</sub> g ce 2no g ce n<sub>2</sub>o g ce h<sub>2</sub>o g nonumber has been experimentally determined to be rate

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web the absorbance of a particular wavelength of light by a solution 2 in this experiment you will conduct the reaction

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web 1 use table 1 to determine the reaction order for hcl 2 use table 2 to determine the reaction order for na<sub>2</sub>s<sub>2</sub>o<sub>3</sub> remember you want to see what happens to the reaction

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