

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. metals _____
- b. metal ions _____
- c. anodes _____
- d. cathodes _____
- e. half-cell reactions: Zn _____
- f. displacement overall reactions _____

- g. half-cell reduction potentials _____
- h. E_{cell} _____

i. Write overall cell equation: $\text{Zn} + \text{Cu}^{2+} \rightarrow \text{Zn}^{2+} + \text{Cu}$ (10 marks)
Write displacement cell half-reactions for copper & silver

Voltaic Cells Study Guide

Terry C. Jones



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