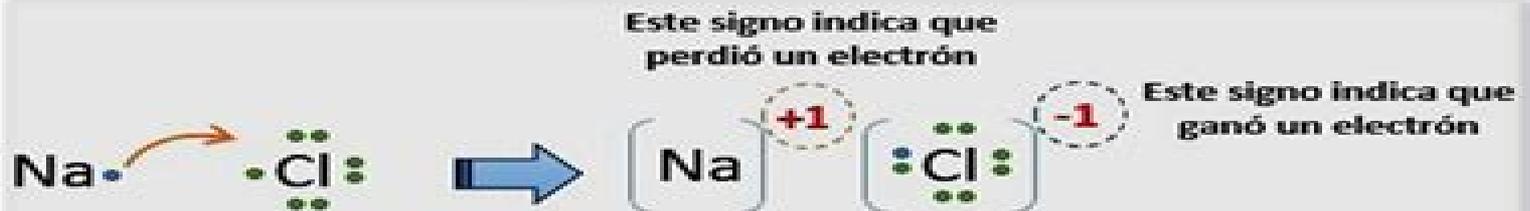


REGLA DEL OCTETO



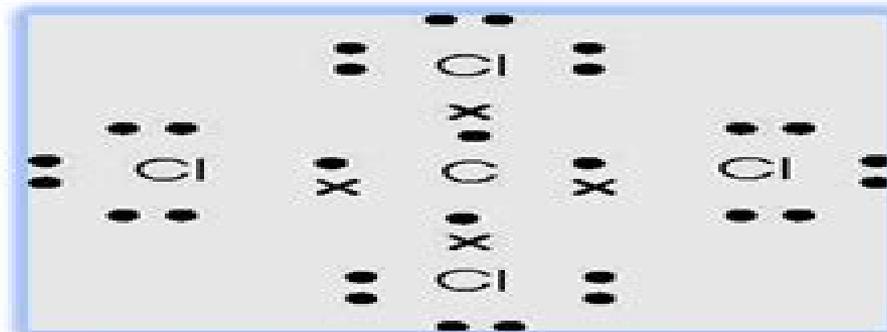
En la unión de estos átomos, el sodio (Na) transfiere su único electrón de valencia al cloro para que pueda completar su octeto



El cloro y azufre comparten sus electrones para completar su octeto (8 electrones de valencia)

REPRESENTACIÓN DE ENLACES CON LA ESTRUCTURA DE LEWIS

Los electrones de los orbitales externos se indican por medio de puntos o cruces alrededor del elemento.



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