

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. anode _____
- b. cathode _____
- c. anion _____
- d. cation _____
- e. direction of electron flow _____
- f. direction of ion movement _____
- g. half-cell reduction equation _____
- h. cell _____

i. Write a word and phrase which describes the cell's ability to convert chemical energy to electrical energy and the equation for the cell's operation with the appropriate half-cell reactions for copper & zinc!

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