

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are the characteristic equations for acid-base chemistry? _____
- (A) pH, $\log K_a$, $\log K_b$, $\log K_w$ (B) pH, $\log K_a$, $\log K_b$, $\log K_w$, $\log K_{sp}$
- (C) pH, $\log K_a$, $\log K_b$, $\log K_w$, $\log K_{sp}$, $\log K_{eq}$ (D) pH, $\log K_a$, $\log K_b$, $\log K_w$, $\log K_{sp}$, $\log K_{eq}$, $\log K_{fwd}$

2. Which of the following is NOT a characteristic of acid-base chemistry? _____
- (A) pH is a measure of the concentration of H^+ ions in a solution. (B) pH is a measure of the concentration of OH^- ions in a solution.
- (C) pH is a measure of the concentration of H^+ ions in a solution. (D) pH is a measure of the concentration of OH^- ions in a solution.

3. In a 0.1 M solution of acetic acid, which of the following is the most abundant species? _____
- (A) CH_3COOH (B) CH_3COO^- (C) H^+ (D) OH^-

4. In a 0.1 M solution of acetic acid, which of the following is the most abundant species? _____
- (A) CH_3COOH (B) CH_3COO^- (C) H^+ (D) OH^-

5. Which of the following is NOT a characteristic of acid-base chemistry? _____
- (A) pH is a measure of the concentration of H^+ ions in a solution. (B) pH is a measure of the concentration of OH^- ions in a solution.
- (C) pH is a measure of the concentration of H^+ ions in a solution. (D) pH is a measure of the concentration of OH^- ions in a solution.

6. Which of the following is NOT a characteristic of acid-base chemistry? _____
- (A) pH is a measure of the concentration of H^+ ions in a solution. (B) pH is a measure of the concentration of OH^- ions in a solution.
- (C) pH is a measure of the concentration of H^+ ions in a solution. (D) pH is a measure of the concentration of OH^- ions in a solution.

7. In a 0.1 M solution of acetic acid, which of the following is the most abundant species? _____
- (A) CH_3COOH (B) CH_3COO^- (C) H^+ (D) OH^-

8. In a 0.1 M solution of acetic acid, which of the following is the most abundant species? _____
- (A) CH_3COOH (B) CH_3COO^- (C) H^+ (D) OH^-

9. In a 0.1 M solution of acetic acid, which of the following is the most abundant species? _____
- (A) CH_3COOH (B) CH_3COO^- (C) H^+ (D) OH^-

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Lei Shi



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