

Voltaic Cells

Label the parts for the following cell:



By connecting the two electrodes, electrons can be transferred and electricity is produced.

- a. anode _____
- b. cathode _____
- c. anion _____
- d. cation _____
- e. salt bridge _____
- f. displacement reaction _____

- g. silver-silver chloride electrode _____
- h. Zn _____

i. Write the half-cell and overall cell reactions for the cell represented by the notation $\text{Zn} | \text{Zn}^{2+} || \text{Cu}^{2+} | \text{Cu}$.

Voltaic Cells Study Guide

K Payea



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web answer these multiple choice questions on topics such as an example of a call center agent becoming angry after taking a call from an angry customer and the term for an agent s ability to

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