

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. metals
- b. metal ions
- c. metals
- d. metal ions
- e. zinc atoms (oxidation) lose electrons
- f. copper ions (reduction) gain electrons

- g. zinc atoms (oxidation) lose electrons
- h. zinc

i. Both metals will give electrons to the solution. The solution will be reduced by either copper ions or zinc ions depending on which is more reactive.

Voltaic Cells Study Guide

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Voltaic Cells Study Guide:

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