

Scientific Inquiry Cycle: Review

35+ QUESTIONS

GRADE
08

1. What factor should be considered when designing a local science solution?
 - a. Resources only
 - b. Spreading environmental issues
 - c. Time constraints
 - d. Member queries
2. What should you do if encountering one part of a scientific inquiry problem is not working?
 - a. Ignore the non-problem
 - b. Repeat the task offensively
 - c. Stop the problem completely
 - d. Fix the problem without thinking
3. What is/are is a local problem that science can help solve?
 - a. Waste management in the community
 - b. Access to clean water
 - c. Access to electricity
 - d. Pollution from factories
4. How does identifying a solution help in the scientific process?
 - a. Shows the problem has been solved
 - b. Provides evidence from using the idea
 - c. Helps others understand and build on it
 - d. Adds unnecessary steps
5. What should be done if a scientific solution is unclear?
 - a. Modify or abandon it for safety
 - b. Use it anyway
 - c. Take it slowly
 - d. Ignore safety and focus on speed
6. Which is an example of a consensus?
 - a. Deciding without any input
 - b. Unanimous input to finish the project
 - c. No input or resources
 - d. Choosing majority feedback for input

Name: _____

Scientific Inquiry Cycle: Review

Science helps communities solve problems by applying knowledge and creating things that "solve" with a local issue such as water quality or waste management using science to make or improve solutions can have a positive impact. Understanding the problem fully, including its causes and effects, is the first step in finding an effective solution.

Designing solutions requires considering various conditions, such as budget, time, and safety concerns. Community members have a part of a solution might face challenges, however, and it is important to weigh these trade-offs carefully. For example, creating materials for a house that may affect its cost and ease of use, cannot influence the effectiveness of the solution.

After identifying a solution, determining how it works, what conditions affected it, and any compromises that have already been or should involve on the way. Communication, using design models, and reflecting on the results of the solution are important steps. The process of testing, sharing, and evaluating locally encourages responsible decision-making and helps communities thrive.

01. Mark the best answer for each question.

1. What is the first important step when working on a local problem?
 - a. Designing a solution
 - b. Spreading the most expensive materials
 - c. Ignoring safety concerns
 - d. Understanding the problem fully
2. What does determining trade-offs in a solution help with?
 - a. Finding the problem of the solution
 - b. Evaluating the costs, benefits, and safety
 - c. Showing how to avoid
 - d. Determining what the solution has to solve
3. Which is an example of a trade-off?
 - a. Finding a solution with no cost
 - b. Spending money on a solution
 - c. The time and resources needed to solve a problem
 - d. Using unsafe materials
4. Why is compromise important?
 - a. To avoid not being in agreement
 - b. To make ideas stronger
 - c. To finish the problem by combining
 - d. To avoid discussing trade-offs



02. Mark if the statement is True (T) or False (F).

1. A solution made after one failed attempt is always better than the first attempt.

Answer for each question.

1. How does identifying a solution help in the scientific process?

2. Why must trade-offs be considered in designing a solution?

3. What might happen when creating a solution that

4. How might science solutions be implemented?

5. Why are trade-offs essential in scientific understanding?

ANSWER KEY
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BC

2nd Semester Inquiry Review Answer Key

JS Bruner



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web a masterpiece one of the best studies on man s desperation helped by wonderful performances and and a stirring poignant script with no wasted lines you will raise goosebumps with scenes like the tango scene you will laugh you will cry you will feel empathy sorrow anger

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web jan 26 2016 the solubility product or ksp is a way of showing how effective a chemical is at dissolving into an aqueous solution the higher the ksp the more the chemical will dissolve into that solvent for this lab calcium hydroxide is created then ionized $Ca(OH)_2$ when calcium is put in water reacts as follows $Ca(OH)_2 + H_2O \rightleftharpoons Ca^{2+} + 2OH^-$

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web k sp of calcium hydroxide 1 author j m mccormick last update ocober 13 2013 introduction an equilibrium constant k is related to ΔG for a process through eqn 1 and as long as ΔH and ΔS are independent of temperature eqn 1 can be combined

with eqn 2 to determine Δh and Δs from the temperature dependence of k

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web your primary objective in this experiment is to test a saturated solution of calcium hydroxide and use your observations and measurements to calculate the ksp of the compound you will do this by titrating the prepared ca oh 2 solution with a standard hydrochloric acid solution

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide ca oh 2 by measuring the concentration of ca oh 2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of ca oh is represented in equation form as shown below ca oh s ca aq 2oh aq the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 à ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is ca oh 2 ca2 2 oh

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molarity 0.050 M volume of filtered Ca(OH)₂ solution ml 5.00 5.00 3.71 13.75 equivalence point volume ml OH⁻ in saturated Ca(OH)₂ M Ca²⁺ in saturated Ca(OH)₂ M K_{sp} of Ca(OH)₂ average K_{sp} of Ca(OH)₂

chemistry 1210 lab 11 experiment 11 determining the K_{sp} of calcium - Sep 08 2023

web the chemical equation used was Ca(OH)₂ (s) ⇌ Ca²⁺ (aq) + 2OH⁻ (aq) the solubility product was determined to be K_{sp} Ca(OH)₂ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

18.1 solubility product constant K_{sp} chemistry libretxts - Feb 01 2023

web jul 12 2023 because the concentration of a pure solid such as Ca₃(PO₄)₂ is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore $K_{sp} = [Ca^{2+}]^3 [PO_4^{3-}]^2$

post lab report 8 determining the K_{sp} of calcium hydroxide - Jul 06 2023

web lab report 8 determination of calcium hydroxide the purpose of this was to titrate a saturated Ca(OH)₂ solution with a standard HCl solution and find the K_{sp} or the OH⁻ concentration was found and then the concentration was found and used for this experiment a titration of calcium hydroxide with hydrochloric acid

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web question determining the K_{sp} of calcium hydroxide Ca(OH)₂ (s) ⇌ Ca²⁺ (aq) + 2OH⁻ (aq) for this experiment the titration of calcium hydroxide with hydrochloric acid was investigated determining the K_{sp} of calcium hydroxide Ca(OH)₂ (s) ⇌ Ca²⁺ (aq) + 2OH⁻ (aq) for this experiment the titration of calcium hydroxide with hydrochloric acid

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web question chem 182 experiment 8 determining the K_{sp} of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of Ca(OH)₂ is represented in equation form solution of calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution as shown below

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest K_{sp} value to the theoretical K_{sp} for Ca(OH)₂ was the determination through pH method 2 introduction calcium hydroxide is a strong base soluble in water

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web plasma membrane homework symbiosis practice worksheet benzoin condensation grade a determining the K_{sp} of calcium hydroxide after reading the lab complete items a b c and d title purpose chemicals and equipment and summary of procedure as described on page 10 of exp 1 on an 8.5 x 11 sheet of paper

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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