

Chemistry Study Guide

Periodic Table of the Elements

1	2											3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18																																																																								
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3	4											Na	Mg	Al	Si	P	S	Cl	Ar																																																																																
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	Cs	Ba											Rn																																																			
55	56											Fr	Ra	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Fl	Uup	Lv	Uus	Uuo																																																																							
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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

Matter:

Anything that has **mass** and **occupies space** 🌟🌍

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the **smallest, indivisible particle** 🔬👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

General Chemistry W Revised Study Guide

Judd E. Hollander



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Table of Contents General Chemistry W Revised Study Guide

1. Understanding the eBook General Chemistry W Revised Study Guide
 - The Rise of Digital Reading General Chemistry W Revised Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying General Chemistry W Revised Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an General Chemistry W Revised Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from General Chemistry W Revised Study Guide
 - Personalized Recommendations
 - General Chemistry W Revised Study Guide User Reviews and Ratings
 - General Chemistry W Revised Study Guide and Bestseller Lists
5. Accessing General Chemistry W Revised Study Guide Free and Paid eBooks
 - General Chemistry W Revised Study Guide Public Domain eBooks
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 - General Chemistry W Revised Study Guide Budget-Friendly Options

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 - General Chemistry W Revised Study Guide Compatibility with Devices
 - General Chemistry W Revised Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of General Chemistry W Revised Study Guide
 - Highlighting and Note-Taking General Chemistry W Revised Study Guide
 - Interactive Elements General Chemistry W Revised Study Guide
8. Staying Engaged with General Chemistry W Revised Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers General Chemistry W Revised Study Guide
9. Balancing eBooks and Physical Books General Chemistry W Revised Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection General Chemistry W Revised Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine General Chemistry W Revised Study Guide
 - Setting Reading Goals General Chemistry W Revised Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of General Chemistry W Revised Study Guide
 - Fact-Checking eBook Content of General Chemistry W Revised Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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- Interactive and Gamified eBooks

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