

Chemistry Study Guide

Periodic Table of the Elements

1	2											3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
H	He											B	C	N	O	F	Ne										
Li	Be											Al	Si	P	S	Cl	Ar										
Na	Mg	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr										
K	Ca	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe										
Rb	Sr		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn										
Cs	Ba		Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Fl	Uup	Lv	Uus	Uuo										
Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Fl	Uup	Lv	Uus	Uuo										
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu											
		Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr											

Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

Matter:

Anything that has **mass** and **occupies space** 🌟🌍

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the **smallest, indivisible particle** 🔬👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

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Table of Contents General Chemistry Study Guide

1. Understanding the eBook General Chemistry Study Guide
 - The Rise of Digital Reading General Chemistry Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying General Chemistry Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an General Chemistry Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from General Chemistry Study Guide
 - Personalized Recommendations
 - General Chemistry Study Guide User Reviews and Ratings

- General Chemistry Study Guide and Bestseller Lists
- 5. Accessing General Chemistry Study Guide Free and Paid eBooks
 - General Chemistry Study Guide Public Domain eBooks
 - General Chemistry Study Guide eBook Subscription Services
 - General Chemistry Study Guide Budget-Friendly Options
- 6. Navigating General Chemistry Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - General Chemistry Study Guide Compatibility with Devices
 - General Chemistry Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of General Chemistry Study Guide
 - Highlighting and Note-Taking General Chemistry Study Guide
 - Interactive Elements General Chemistry Study Guide
- 8. Staying Engaged with General Chemistry Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers General Chemistry Study Guide
- 9. Balancing eBooks and Physical Books General Chemistry Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection General Chemistry Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine General Chemistry Study Guide
 - Setting Reading Goals General Chemistry Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of General Chemistry Study Guide
 - Fact-Checking eBook Content of General Chemistry Study Guide
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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