

# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe the components and direction of electron flow.

- a. anode \_\_\_\_\_
- b. cathode \_\_\_\_\_
- c. anion \_\_\_\_\_
- d. cation \_\_\_\_\_
- e. direction of electron flow \_\_\_\_\_
- f. direction of ion movement \_\_\_\_\_
- g. half-cell reduction equation \_\_\_\_\_
- h. cell \_\_\_\_\_
- i. Write overall cell equation with phases.  $\text{cell} = \text{anode} | \text{anode solution} || \text{cathode solution} | \text{cathode}$  represent the entire experiment with half-cells separated by a salt bridge.

\_\_\_\_\_

# Voltaic Cells Study Guide

**D Keegan**



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