

**BAGONG SILANG NATIONAL HIGH SCHOOL**  
**1<sup>ST</sup> QUARTER EXAMINATION**  
**PHYSICAL SCIENCE**

NAME: \_\_\_\_\_  
 GRADE: \_\_\_\_\_

DATE: \_\_\_\_\_  
 SCORE: \_\_\_\_\_

Directions: Write the letter of the correct answer on the space provided.

- \_\_\_\_\_ 1. It is also known as nuclear fusion and the formation of new nuclei actions in the early stages of development of the universe.  
 a. Nucleosynthesis                      b. S-Process                      c. R-Process                      d. Singularity
- \_\_\_\_\_ 2-4. There are two categories of star, the average star and massive star. The life cycle of these two stars are common from 1<sup>st</sup> to 3<sup>rd</sup> stages namely nebula, protostar, and main sequence star. What will be the next stages in the life of an average star until it reaches its death?  
 a. Red super giant → Supernova → Black hole → neutron star 0  
 b. Red giant → white dwarf → black dwarf 3  
 c. Red giant → white dwarf → black hole 2  
 d. Red supergiant → white dwarf → black dwarf 1
- \_\_\_\_\_ 5. He synthesized element with atomic number 43 using a linear particle accelerator.  
 a. Ernest Rutherford    b. Ernest Lawrence    c. Dmitri Mendeleev    d. John Dalton
- \_\_\_\_\_ 6. The following are light elements EXCEPT;  
 a. helium                      b. hydrogen                      c. lithium                      d. iron
- \_\_\_\_\_ 7. Which of the following molecule is nonpolar?  
 a. NaCl                      b. HCl                      c. CO<sub>2</sub>                      d. NH<sub>3</sub>
- \_\_\_\_\_ 8. What type of chemical bond holds the atoms of water molecule together?  
 a. Hydrogen bond    b. Ionic bond    c. Polar covalent bond    d. Nonpolar covalent bond
- \_\_\_\_\_ 9. Which of the following IMFAs is considered as the weakest?  
 a. H-bonding    b. Ion-dipole    c. Dipole-dipole    d. London forces
- \_\_\_\_\_ 10. Which of the following statements is TRUE about nonpolar molecules?  
 a. high boiling point    b. high melting point    c. low surface tension    d. low vapor pressure
- \_\_\_\_\_ 11. Forces that are present between an ion and a polar molecule  
 a. ion-dipole    b. dipole-dipole    c. dispersion forces    d. hydrogen bonding
- \_\_\_\_\_ 12. H-bonding forms when the substances involved are polar and there are \_\_\_\_\_.  
 a. H-atoms of a molecule interact with to O, N, F of other molecules.  
 b. C-atoms attached to O, N, F.  
 c. central atoms with O, N, F as attached atoms.  
 d. unshared pair of electrons in the central atom.
- \_\_\_\_\_ 13. The amount of energy required to stretch or increase the surface of a liquid by a unit area  
 a. specific heat    b. surface tension    c. vapor pressure    d. heat of vaporization
- \_\_\_\_\_ 14. Liquids can form spherical elastic film to minimize surface area. What intermolecular force is responsible for the formation of this film in water?  
 a. H-bonding    b. ion-induced dipole    c. dipole-induced dipole    d. London dispersion force
- \_\_\_\_\_ 15. Which group of biomolecules carries and passes on the hereditary information of the organisms?  
 a. carbohydrates    b. lipids    c. nucleic acids    d. proteins
- \_\_\_\_\_ 16. What are the elements that make up carbohydrates?  
 a. C and H    b. C, H and N    c. C, H and O    d. C, H, O and N
- \_\_\_\_\_ 17-19. Is carbohydrates one of the most important biological macromolecules needed by a human body?  
 a. Yes, because it is the primary source of energy. 3  
 b. Yes, because it can be a storage of energy for a long period of time. 1  
 c. Yes, because it can be found in rice, cereals, fruits and pastas. 2  
 d. Yes, because it carries genetic information. 0
- \_\_\_\_\_ 20. The measure of how fast is the change in the concentration of the reactants or products in a chemical reaction.  
 a. Activation Energy    b. Activation Rate    c. Chemical Reaction    d. Concentration
- \_\_\_\_\_ 21. A chemical change that occurs when two or more substances combine to form a new substance.  
 a. Activation Energy    b. Chemical Kinetics    c. Chemical Reaction    d. Product

# Section 2review Answer Key Physical Science

**A Loxley**



## Section 2 review Answer Key Physical Science:

CliffsNotes Praxis II: Middle School Science (0439) Glen Moulton, 2013-05-21 Your complete guide to a higher score on Praxis II Middle School Science The Praxis II Middle School Science 0439 exam is designed to measure the knowledge and competencies necessary for a beginning teacher of middle school science The 2 hour Praxis II Middle School Science 0439 exam consists of three constructed response essays and 90 multiple choice questions divided into the following content categories scientific methodology basic principles of science physical sciences life sciences earth space sciences and science technology society In CliffsNotese Praxis II Middle School Science two practice tests with complete answers and explanations help you pinpoint areas for further study while reviews and exercises address all of the test topics you ll encounter on exam day Plus proven test taking strategies help you score higher Two full length practice tests Subject reviews of every topic covered on the test Practice questions for every subject review If you re an aspiring teacher looking to take the Praxis II Middle School Science exam CliffsNotes is your ticket to scoring high at exam time **CLEP® General Exams Book + Online, 9th Ed.** Stu Schwartz, Laurie Callihan, Scott Dittloff, 2018-01-30 CLEP General Exams Book Online Practice Tests Helps Students Get the College Credits They Deserve 9th Edition In 2017 CLEP marks 50 years as the most widely trusted credit by exam program in the U S CLEP exams help students fast track their college degree saving them time and possibly thousands in tuition costs Perfect for adults returning to college military service members high school or home schooled students REA s CLEP test preps provide students with the tools they need to pass their CLEP exams and get the college credits they deserve REA s new 9th edition of the CLEP General Exams bundles complete test prep for the four CLEP general exams College Mathematics Humanities Natural Sciences Social Sciences History that satisfy typical first year general education requirements These are the courses for which most community and military friendly colleges will award CLEP credit About REA s Prep Complete test prep for the 4 CLEP general exams College Mathematics Humanities Natural Sciences and Social Sciences History Great consumer value only 34 95 4 comprehensive review sections 1 for each CLEP exam 4 online diagnostic tests 1 for each CLEP exam 8 full length practice tests 2 for each CLEP exam Online diagnostic and practice tests feature instant scoring timed testing diagnostic feedback and detailed answers *Addison-Wesley*

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