

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance
 (B) volume, molarity, pH, amount of substance
 (C) volume, molarity, pH, concentration of ions
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A) $\text{pH} = \frac{1}{10} \log \frac{1}{1000}$ (B) $\text{pH} = \frac{1}{10} \log 1000$

(C) $\text{pH} = \frac{1}{10} \log \frac{1}{1000}$ (D) $\text{pH} = \frac{1}{10} \log 1000$ (E) amount of substance

2. An acid-base reaction represents a proton transfer from a stronger acid to a weaker base. Which of the following is not correct, and why? (correct)
- (A) $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$ (B) $\text{H}_2\text{O} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$ (C) $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{O} + \text{HCl}$ (D) $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$



3. An acid-base reaction involves the transfer of a proton from the stronger acid to the weaker base. Which of the following is not correct? (correct)
- (A) $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$ (B) $\text{H}_2\text{O} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$ (C) $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{O} + \text{HCl}$ (D) $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$

4. Which of the following is not a property of the following equation? (correct)



5. A solution of acetic acid and its conjugate base is prepared by mixing acetic acid and sodium acetate. If a 100 mL of acetic acid is mixed with 100 mL of 0.1 M sodium acetate, what is the pH of the resulting solution? (correct)



6. An acid-base equilibrium, $\text{CH}_3\text{COOH} + \text{H}_2\text{O} \rightleftharpoons \text{CH}_3\text{COO}^- + \text{H}_3\text{O}^+$, is established in a solution of CH_3COOH . After this reaction is completed, which of the following is not a species in the reaction mixture? (correct)



Multiple Choice Stoichiometry Test

Brendan G. Carr



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